



**District Administration, Zilla Panchayat Dharwad**  
DDPI Office, Department of School Education, Dharwad



**Subject: Science**

Student friendly Study Resource Material for  
the Success in S.S.L.C. Examination

**Coordination**

DDPI (Admin) and (Development), Education Officers,  
District Planning and Deputy Coordinating Officer, BEOs, Subject  
Inspectors and Science Subject Forum.



District Administration Dharwad  
School Education and Literacy Department Dharwad

## Vignana-Vidya Vishwas

### SSLC STUDENT PASSING PACKAGE

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Government of Karnataka



**Smt. Divyaprabhu G.R.J. IAS**

District Administration Dharwad  
Honorable Deputy Commissioner and  
District Magistrate, Dharwad District  
Dharwad.

Dear Students,

I hope you read this message with attention and face the SSLC examination with confidence, achieving excellent results.

The educational experts of Dharwad district have come together and initiated a program called "Mission Vidyakashi," which aims at improving the examination results and preparing all SSLC students to achieve their desired goals. Each of you has the potential, capability, and dreams to achieve success, and the supportive environment will help you realize them. The SSLC examination is a significant milestone for fulfilling your dreams and achieving your future goals. To score well in this examination, consistent study is essential.

You might often wonder, "How should our daily study routine be structured?" To answer this, Mahatma Gandhi once said, "We must think about what we read, analyze it, and make it an integral part of our daily life." Keeping this thought in mind, understand and memorize what your teachers teach you daily and reinforce it through regular revision.

This effort will help shape the success of your future endeavors. The education department, with the support of teachers, has prepared additional learning materials that are easy to understand for all students appearing for the SSLC examination. Students should make good use of these resources to achieve excellent results.

With this in mind, we hope that the district of Dharwad will achieve 100% success at the state level as well.

Wishing you all the best for your future endeavors!

**Smt. Divyaprabhu G.R.J.**

District Commissioner and District Magistrate,

Dharwad District, Dharwad

Government of Karnataka

District Administration, Zilla Panchayat, and Department of School Education Dharwad"



## "Preface"

A school aims to provide quality education to shape the future effectively. The school, with its various initiatives, focuses on nurturing the unique qualities of each child and adapting teaching methods according to their needs to achieve excellence in education. Students' academic performance is evaluated through exams.

The SSLC exam is a significant milestone in a student's life. It is essential for every student to succeed in this examination. With the goal of helping all students achieve excellent marks and excel in their future educational endeavors, teachers have prepared supplementary materials for all subjects to support students. These materials are expected to be highly beneficial in preparing for the upcoming exams.

Considering the individual learning pace of each student, teachers have structured these supplementary materials as a model, guiding them to achieve excellent results. It is hoped that students will make the best use of these resources, face the exams with confidence, and achieve success.

S. SKeladimath DDPI

(Admin)

## "Physics"

"There are currently 4 chapters in Physics, with questions totaling 28 marks. This study guide is designed to help you easily score at least more than 15 marks.

Chapters:

1. Electricity
2. Magnetic Effects of Electric Current
3. Light, Reflection, and Refraction
4. The Human Eye and the Colorful World

Chapter 1 – Electricity

"Four multiple-choice answers are provided for the following questions or incomplete statements. Choose the correct answer from them and write the complete answer along with the corresponding option letter."

1. The metal used in electric bulbs.

- A) Manganese
- B) Tungsten
- C) Nickel
- D) Chromium

**Answer:** B) Tungsten

2. The device used to change the resistance in an electric circuit:

- a) Voltmeter
- b) Ammeter
- c) Galvanometer
- d) Rheostat

**Answer:** d) Rheostat

3. The device used to generate electricity:

- a) Galvanometer
- b) Generator
- c) Ammeter
- d) Electric Motor

Answer: b) Generator

4. The correct formula showing the relationship between potential difference, electric current, and resistance in an electric circuit:

- a)  $I = R/V$
- b)  $I = VR$
- c)  $V = I/R$
- d)  $R = V/I$

Answer: c)  $V = I/R$

"Answer the following questions in one sentence. (1 mark questions)"

1. What is the international unit of potential difference?

Answer: Volt

2. Name the device used to measure electric current.

Answer: Ammeter

3. Define electric potential difference.

Answer: The electric potential difference is the amount of work needed to move a unit charge from one point to another in an electric circuit.

4. How is an ammeter connected in an electric circuit?

Answer: It is connected in series.

5. How is a voltmeter connected in a circuit?

Answer: It is connected in parallel

6. What is a resistor?

Answer: A conductor with significant resistance is called a resistor.

7. What is the international unit of resistivity?

Answer: Ohm-meter

8. Why are alloys used in heating appliances?

Answer: Alloys have high resistivity and do not burn out at high temperatures, which is why they are used in heating appliances.

9. What are some devices that operate under the principle of heating effect?

Answer: Iron box, electric oven, electric heater, electric kettle, hair dryer, etc.

10. What is the practical unit of electrical energy?

Answer: Kilowatt-hour

Answer the following questions in 2-3 sentences (2 marks)

1. State Ohm's Law.

Answer: At a constant temperature, the potential difference  $(V)$  across a conductor is directly proportional to the current  $(I)$  flowing through it. The relationship is given by  $(V = IR)$ .

2. What factors does the resistance of a conductor depend on?

Answer: The resistance of a conductor depends on 1) the length of the conductor, 2) the cross-sectional area of the conductor, and 3) the material's inherent properties.

3. State Joule's Law of Heating.

Answer: The heat produced in a resistor is:

- 1) Directly proportional to the square of the electric current passing through it.
- 2) Directly proportional to the resistance of the conductor.
- 3) Directly proportional to the time for which the electric current flows through the conductor.

The formula is  $H = I^2 R t$

4. What is a fuse? What is its function?

Answer: A fuse is a wire made of a metal or alloy with a suitable melting point. It protects electrical devices by preventing excessive current from flowing through the circuit, melting when the current exceeds a safe level.

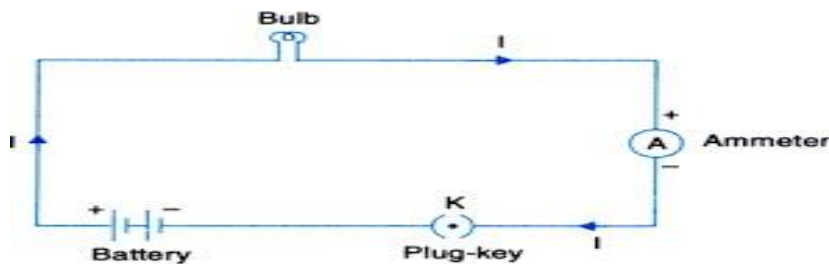
5. How does a fuse work?

Answer: When an electric current higher than the specified limit flows through the circuit, the temperature of the fuse wire increases. As a result, the fuse wire melts, breaking the circuit. This protects the electrical appliances in the house from damage.

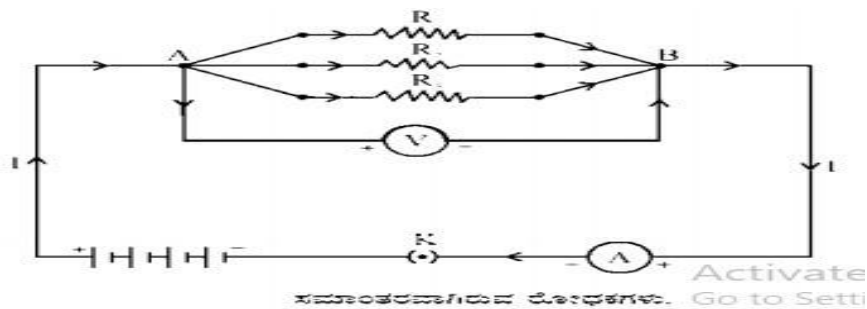
6. What is electric power? What is its international unit?

Answer: Electric power is the rate at which electrical energy is consumed or used. Its international unit is the watt.

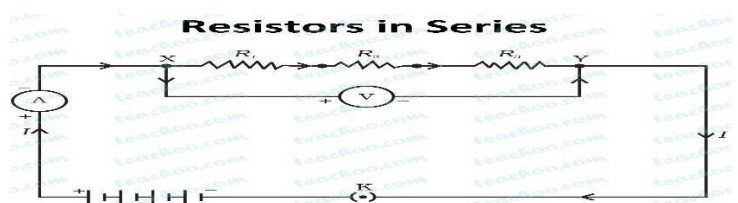
7. "Draw a diagram of a simple electric circuit."



8. "Draw a diagram of resistors connected in parallel, and label the ammeter, voltmeter, and switch."



9. Draw a diagram of resistors connected in series, and label the components."





10. A refrigerator with a power rating of 400 W is used for 8 hours a day. If the cost per 1 kWh is ₹3, what is the total energy cost for 30 days?

Answer:

Total energy consumed by the refrigerator in 30 days:

At ₹3 per kWh, the total cost for 30 days:

" $400\text{W} \times 8\text{hours} \times 30\text{days} = 96,000\text{Wh} = 96\text{kWh}$ " "But for 1 kWh, it is ₹3 for 30 days."

"Total energy used by the refrigerator =  $96\text{kWh} \times ₹3\text{per kWh} = ₹288.00$ "

## "Chapter 2 - The Magnetic Effects of Electric Current"

"For the following questions or incomplete statements, four alternative answers are provided. Choose the appropriate answer from them and write the complete answer with the corresponding letter."

1. In Fleming's Left-Hand Rule, the direction indicated by the middle finger is:

- a) Magnetic Field
- b) Electric Current
- c) Movement of the Conductor
- d) Induced Electric Current

Answer: Electric Current

2. A conductor wire wound in the form of a coil is called:

- a) Coil
- b) Solenoid
- c) Commutator
- d) Dynamo

Answer: Coil

3. The magnetic field inside a long straight solenoid with an electric current:

- a) Is uniform at all points
- b) Is zero
- c) Decreases as you move toward the ends
- d) Increases as you move toward the ends

Answer: Is uniform at all points

4. The magnetic field lines inside a solenoid are parallel straight lines because the magnetic field inside a solenoid:

- a) Is very strong
- b) Is uniform
- c) Is zero
- d) Contains electric current

Answer: Is uniform

5. In Fleming's Right-Hand Rule, if the thumb indicates the direction of the electric current, the remaining fingers show the direction of:

- a) Induced Electric Current
- b) Magnetic Field
- c) Movement of the Conductor
- d) Mechanical Force

Answer: Magnetic Field

"Answer the following questions in a single sentence. (1 mark questions)"

1. Magnetic field lines do not intersect each other. Why?

Answer: At the point of intersection, the compass needle would have to point in two directions simultaneously, which is impossible, so magnetic field lines do not intersect each other.

2. What is meant by "overload"?

Answer: Overload refers to connecting multiple devices to an electric current beyond the capacity of the conductor.

3. Mention two measures to prevent overload in a household electrical system.

Answer: 1) Do not connect multiple appliances to a single outlet (socket).

2) Ensure that live and neutral wires do not come into direct contact.

4. What causes an overload in an electrical system?

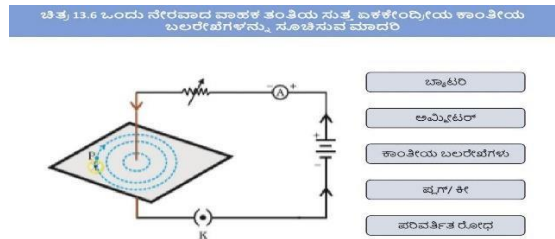
Answer: Overload can be caused by a sudden increase in voltage, connecting multiple appliances to a single outlet (socket), and direct contact between live and neutral wires.

5. How should household electrical appliances be connected?

Answer: Household electrical appliances should be connected in parallel, which helps in saving electricity.

Answer the following questions in 2-3 sentences. (2 mark questions)."

1. Draw a diagram showing concentric circular lines of magnetic force around a



straight current-carrying conductor.

2. What are the functions of an earth wire?

Answer: The earth wire is used for the safety of electrical appliances with metallic surfaces in residential and commercial electrical systems. It channels excess charge flowing through the systems safely to the ground, thereby neutralizing it.

3. Why is it necessary to earth electrical appliances with metallic surfaces in household electrical systems?

Answer: It reduces resistance, ensures that any electric leakage in the appliances is equalized with the earth potential, and provides protection from electric shocks.

4. How do you identify the magnetic field lines around a bar magnet?

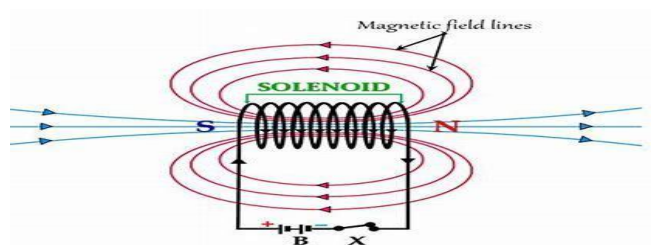
Answer: Place a bar magnet and a compass on a sheet of white paper. Starting from any point, move the compass to different positions, marking the direction with a pencil. Connecting all these points will reveal the magnetic field lines.

5. Write the characteristics of magnetic field lines.

Answer:

- 1) They do not intersect each other.
- 2) They originate from the north pole and end at the south pole.
- 3) The density of lines is higher near the magnetic poles.
- 4) While they appear to move from north to south outside the magnet, they travel from south to north inside the magnet.

6. Draw a solenoid and label its parts.



7. What is a solenoid and what is its use?

Answer: A solenoid is a cylindrical shape formed by tightly wound coils of copper wire carrying an electric current. Solenoids are used to generate electromagnetic fields, control electrical circuits (turning them on or off), and in hydraulic machines.

8. Define Fleming's Right-Hand Rule.

Answer: "When you hold a straight conductor with electric current in your right hand, with the thumb, index finger, and middle finger mutually perpendicular, the index finger indicates the direction of the magnetic field, the middle finger indicates the direction of the electric current, and the thumb indicates the direction of the force on the conductor."

9. Define Fleming's Left-Hand Rule.

Answer: "When you hold your left hand with the thumb, index finger, and middle finger mutually perpendicular, the index finger indicates the direction of the magnetic field, the middle finger indicates the direction of the electric current, and the thumb indicates the direction of the force acting on the conductor."

10. What is a short circuit and how can it be prevented?

Answer: A short circuit occurs when live and neutral wires accidentally come into direct contact, or during an overload, causing an interruption in the electric connection. It can be prevented by installing fuses and taking precautions to avoid overloads.

### Chapter-3. Light, Reflection, and Refraction

For the following questions or incomplete statements, four alternative answers are provided. Choose the appropriate answer from them and write the complete answer with the corresponding letter.

1. Which of the following materials is not used in the manufacture of lenses?

- a) Water      b) Glass      c) Plastic      d) Clay

Answer: Plastic

2. The image formed by a concave mirror is virtual, erect, and larger than the object. Thus, the position of the object is:

- a) Between the principal focus and the center of curvature  
b) At the center of curvature  
c) Beyond the center of curvature  
d) Between the pole and the principal focus

Answer: Between the pole and the principal focus

3. What type of mirror is used as a rear-view mirror in vehicles?

- a) Plane  
b) Concave  
c) Convex  
d) Plane or Convex

Answer: Convex

4. When light strikes a smooth surface and returns to the same medium, it is called:

- a) Refraction
- b) Reflection
- c) Dispersion
- d) None of the above

Answer: Reflection

5. The process of focusing or defocusing of light in a lens is known as:

- a) Power
- b) Strength
- c) Magnification
- d) Dispersion

Answer: Power

Answer the following in one sentence (one mark)

1. What is reflection of light?

Answer: Reflection of light is when light strikes a smooth surface and returns to the same medium.

2. Where are concave mirrors used?

Answer: Concave mirrors are used in torches, searchlights, and vehicle headlights to obtain a powerful parallel beam of light, and in shaving mirrors and dental instruments.

3. What is the reason a coin placed in water appears to be raised?

Answer: The reason is light refraction.

4. What is the refractive index?

Answer: The refractive index is the measure of the change in direction that occurs when light passes from one medium to another, relative to the two media.

5. What is the international unit of power of a lens?

Answer: Diopter (D)

Answer the following 2-3 sentences (2 marks)

1. What is refraction of light?

Answer: Refraction of light occurs when light changes direction as it passes from one medium with a different density to another medium. This bending of light is known as refraction.

2. What are the characteristics of the image formed by a plane mirror?

Answer:

1. The images formed by a plane mirror are always virtual and erect.
2. The size of the image is equal to the size of the object.
3. The image is located as far behind the mirror as the object is in front of it.

3. State the laws of reflection of light.

Answer:

1. The incident ray, the point of incidence, and the reflected ray all lie in the same plane.
2. The angle of incidence is equal to the angle of reflection.



4. State the laws of refraction of light.

Answer:

1. The incident ray, the refracted ray, and the normal at the point of incidence all lie in the same plane.

2. For a given pair of media and a specific color of light, the ratio of the sines of the angle of incidence and the angle of refraction is constant. Refractive index =  $\frac{\sin i}{\sin r}$

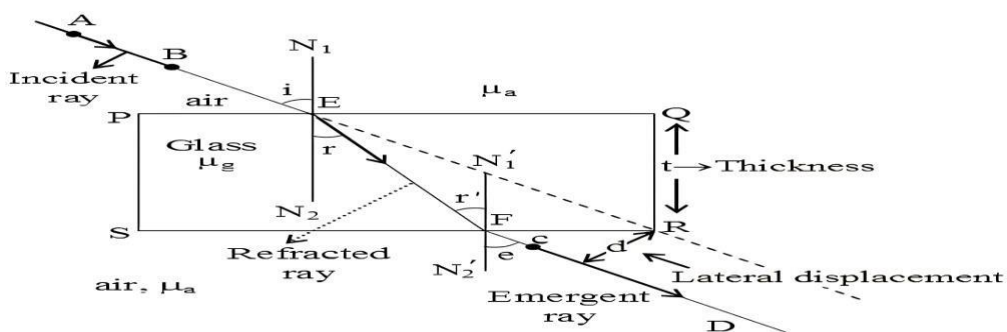
5. State the mirror formula.

Answer: The mirror formula relates the object distance ( $u$ ), the image distance ( $v$ ), and the focal length ( $f$ ) of a mirror. The formula is  $\frac{1}{v} + \frac{1}{u} = \frac{1}{f}$

6. What is magnification of a mirror?

Answer: The magnification of a mirror indicates how much larger or smaller an image appears relative to the size of the object. It is expressed as the ratio of the height of the image to the height of the object.

7. Draw a diagram of light refraction through a rectangular glass slab and label its parts.



8. Eyedoctorsometimesprescribe+2.0Dor-2.5Dlenses.Whatdoesthismean?

Answer:Thisindicatesthepowerofcorrectivelensesneededforvisioncorrection.  
+2.0 D means a convex lens is required for farsightedness, while -2.5 D means  
a concave lens is needed for nearsightedness.

#### **Chapter-4:TheHumanEyeandtheColorfulworld**

Forthefollowingquestionsorincompletestatements,fouralternativeanswersare  
provided.Choosetheappropriateanswerfromthemandwritethecompleteanswerwith the  
corresponding letter.

1. Throughwhichpartofthehumaneyedoestlightenter?

- a) Retina b) Cornea c) Iris d) Pupil

Answer: Cornea

2. Theabilityofthehumaneyetoadjusttodifferentfocallengthsiscalled:

- a) Papilla  
b) DistanceVision  
c) Accommodation  
d) BlindSpot

Answer:Accommodation

3. Themimumdistanceatwhichanobjectmustbefromtheeyetobeseenclearlyis:

- a) 17cm
- b) 25cm
- c) 25m
- d) 17m

Answer:25cm

4. The ability to see nearby objects clearly is referred to as:

- a) Nearsightedness
- b) Farsightedness
- c) Cataract
- d) None of the above

Answer:Nearsightedness

5. Nearsightedness can be corrected using:

- a) Convex lens
- b) Concave lens
- c) Plane lens
- d) Convex-concave lens

Answer:Concave lens

6. Farsightedness can be corrected using:

- a) Convex lens
- b) Concave lens
- c) Plano lens
- d) Convex-concave lens

Answer: Convex lens

7. In the dispersion of light, the color that bends the least is:

- a) Blue
- b) Red
- c) Violet
- d) Green

Answer: Red

8. In the dispersion of light, the color that bends the most is:

- a) Blue
- b) Red
- c) Violet
- d) Green

Answer: Violet

9. The first scientist to use a glass prism to study the dispersion of sunlight was:

- a) Albert Einstein
- b) Thomas Alva Edison
- c) Sir Isaac Newton
- d) Alfred Nobel

Answer: Sir Isaac Newton

Answer the following in one sentence (1 Mark)

1. What is a cataract in the eye?

Answer: A cataract is when the retina appears swollen or cloudy, resembling a milk-white film or foggy atmosphere, often seen in older individuals.

2. What are the causes of nearsightedness?

Answer:

1. Excessive curvature of the eye's lens
2. The eyeball being too long from front to back

3. What is farsightedness?

Answer: Farsightedness is a condition where a person can see distant objects clearly but has difficulty seeing nearby objects.

4. What is the cause of farsightedness?

Answer: Farsightedness is usually caused by the lens of the eye being too flat or the eyeball being too short.

5. What is presbyopia?

Answer: Presbyopia is a condition commonly seen in older individuals where the ability of the eye to focus on nearby objects is diminished, making it difficult to see close objects clearly without glasses.

6. What causes presbyopia?

Answer: Presbyopia is caused by the loss of elasticity in the eye's lens or weakening of the ciliary muscles.

7. What is dispersion of light?

Answer: Dispersion of light is the process by which white light is separated into its component colors, each having different wavelengths.

8. What is a rainbow?

Answer: A rainbow is a natural spectrum of light appearing in the sky, created by the dispersion of light through tiny water droplets in the atmosphere after rainfall.

9. Why do stars twinkle?

Answer: Stars twinkle due to the refraction of their light by the Earth's atmosphere, which causes the stars to appear to flicker.

10. Why do planets not twinkle?

Answer: Planets do not twinkle because they are closer to Earth and appear as extended sources of light rather than points of light, reducing the effect of atmospheric distortion.

11. What is the Tyndall effect?

Answer: The Tyndall effect is the scattering of light by particles in a colloid or in fine suspensions, which makes the path of light visible.

Answer the following in 2-3 sentences (2 marks only)

1. Give reasons for the following:

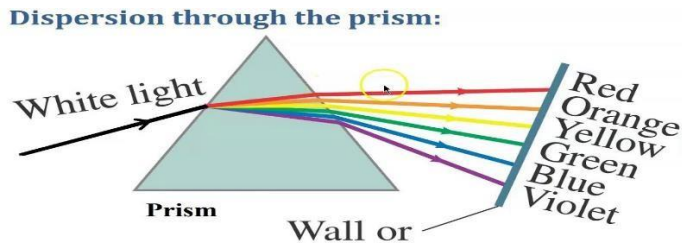
a) Why is the color of the sky blue?

Answer: The sky appears blue because the air molecules and small particles in the atmosphere scatter shorter wavelengths of light (blue) more effectively than longer wavelengths.

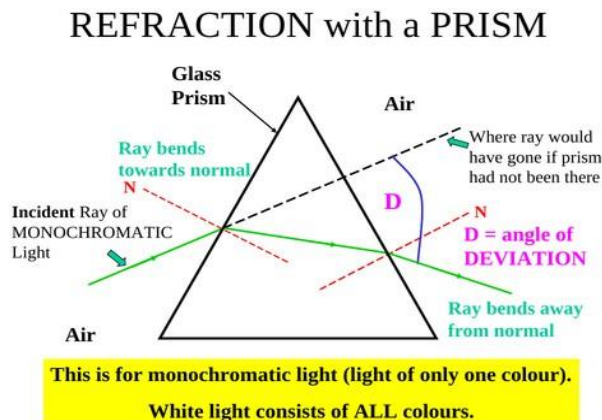
b) Why does the sky appear black to astronauts?

Answer: Astronauts see the sky as black because, at high altitudes beyond the Earth's atmosphere, there is no atmosphere to scatter sunlight, so no scattering of light occurs.

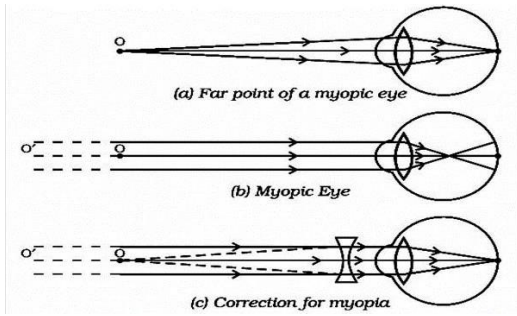
2 Draw and label the dispersion of light in a glass prism.



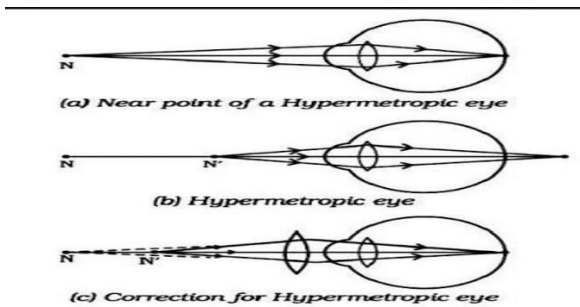
3. Draw and label the dispersion of light through a triangular prism.



4. Draw and label a diagram showing near-sightedness and its correction.



5. Draw and label a diagram showing far-sightedness and its correction.



“Chemistry”



## Chapter Questions:

### Chapter-1 Chemical Reactions and Equations

I. For each of the following questions, four options are provided. Choose the correct answer and write it with the corresponding letter. (1 mark for each question)

- In a chemical equation, the reactants are  
a) Written before the arrow sign  
b) Written after the arrow sign  
c) Written above the arrow sign  
d) Written before the products  
Answer: b) Written after the arrow sign
- The outcome of a chemical reaction is called  
a) Reactant b) Catalyst  
c) Product d) Heat  
Answer: c) Product
- In the equation " $\text{Zinc} + \text{Sulfuric Acid} \rightarrow \text{Zinc Sulfate} + \text{Hydrogen}$ ", the products formed are  
a) Zinc b) Zinc Sulfate c) Hydrogen d) Both b and c  
Answer: d) Both b and c
- In a chemical equation, the symbol(s) represent the state  
a) Solid b) Liquid c) Gas d) Plasma  
Answer: a) Solid
- The symbols (l) & (g) in a chemical equation represent the states  
a) Solid & Gas b) Liquid & Solid  
c) Liquid & Gas d) Gas & Molten  
Answer: c) Liquid & Gas
- The chemical formula for quicklime is  
a) CaO b)  $\text{CaCO}_3$  c)  $\text{CO}_2$  d) CO  
Answer: a) CaO
- The chemical name and formula for slaked lime  
a) Calcium Oxide - CaO  
b) Calcium Hydroxide -  $\text{Ca(OH)}_2$   
c) Calcium Carbonate -  $\text{CaCO}_3$   
d) Carbon Dioxide -  $\text{CO}_2$   
Answer: b) Calcium Hydroxide -  $\text{Ca(OH)}_2$
- The color of ferrous sulfate crystals is

a) Green b) Yellow c) Liquid & Gas d) Gas & Molten

Answer: a) Green

9. The brown fumes produced when lead nitrate is thermally decomposed

- a) Lead Oxide - PbO      b) Oxygen - O<sub>2</sub>      c) Nitrogen Dioxide - NO<sub>2</sub>  
d) Both b & c

Answer: Nitrogen Dioxide - NO<sub>2</sub>

10. Chemical reaction that occurs with the help of sunlight:

- a)  $\text{AgCl} \rightarrow \text{Ag} + \text{Cl}_2$   
b)  $\text{AgBr} \rightarrow \text{Ag} + \text{Br}_2$   
c)  $\text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   
d) Both a & b

Answer: d) Both a & b

II. Answer the following questions in one sentence (1 mark each):

11. What is a chemical combination?

Answer: A chemical reaction in which two or more reactants combine to form a single product.

12. Define chemical decomposition.

Answer: A chemical reaction in which a single reactant breaks down into two or more products.

13. What are the types of chemical reactions?

Answer: Chemical combination, chemical decomposition, chemical displacement, and double displacement.

14. What type of chemical reaction is the combustion of coal?

Answer: Chemical combination.

15. Why does ferrous sulfate lose its green color when heated?

Answer: Due to thermal decomposition, it loses water.

16. What is a thermal decomposition reaction?

Answer: A chemical decomposition reaction that occurs through heating.

17. Why does silver chloride turn grey in sunlight?

Answer: Silver chloride decomposes into silver and chlorine in sunlight.

18. What is an endothermic reaction?

Answer: A chemical reaction in which heat or energy is absorbed.

19. What is oxidation?

Answer: A chemical reaction in which a substance gains oxygen.

20. What is reduction?

Answer: A chemical reaction in which a substance loses oxygen.

21. Why is nitrogen gas passed into chip packets?

Answer: To prevent the chips from becoming soggy.

22. What change occurs when copper undergoes corrosion?

Answer: A green coating appears on the surface of copper.

23. What change occurs when silver undergoes corrosion?

Answer: A black coating appears on the surface of silver.

24. What change occurs when iron undergoes corrosion?

Answer: Iron forms rust.

25. What is rust?

Answer: The aqueous oxide of iron is called rust.

26. What is a precipitate?

Answer: A substance that is formed during a chemical reaction and is insoluble in water is called a precipitate.

III. Answer the following questions in two sentences (2 marks each)

27. Write the difference between chemical combination and chemical decomposition (mention any two points):

Answer:

Chemical Combination

Chemical Decomposition

1. Two or more reactants.

One reactant.

2. One product.  
products

Two or more

3. Example:  $H_2 + O_2 \rightarrow H_2O$   
 $Fe_2O_3 + SO_2 + SO_3$

Example:  $FeSO_4 \rightarrow$

28. Write the difference between oxidation and reduction (mention any two points):

Answer: Oxidation

Reduction

1. The process of losing oxygen.

1) The process of gaining oxygen.

1. The process of gaining electrons.

2) The process of losing electrons

2. The process of gaining hydrogen.

3) The process of losing hydrogen.

3. Example:  $CuO + H_2 \rightarrow Cu + H_2O$

4) Example:  $C + O_2 \rightarrow CO_2$

29. What is a redox reaction? Give an example.

Answer: A reaction in which one reactant undergoes oxidation while another undergoes reduction.

Example:  $CuO + H_2 \rightarrow Cu + H_2O$  or  $ZnO + C \rightarrow Zn + CO$

30. What are corrosion and rancidity?

Answer: Corrosion is the process by which metals are attacked by substances like moisture and acids from their surroundings. Rancidity is the process of oxidation of fats and oils.

31. Write one equation each for a chemical displacement and a double displacement reaction.

Answer:



32. Write the difference between endothermic and exothermic reactions:

Answer:

Endothermic Reaction

Exothermic Reaction

1. The reaction absorbs heat/energy.

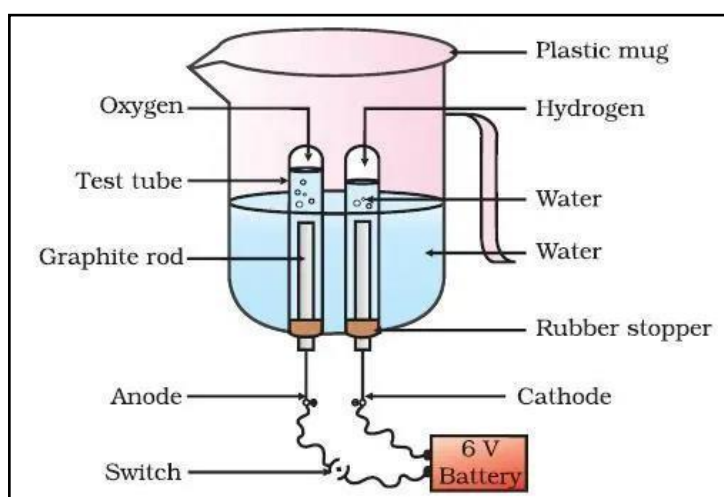
The reaction releases heat/energy

2. Example:  $\text{AgCl} \rightarrow \text{Ag} + \text{Cl}_2$

Example:  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

IV. Answer the following question (3/4 mark each):

33. Draw a neat diagram of the electrolysis of water and label the parts.



Chapter-Specific Questions: Chapter 2 – Acids, Bases, and Salts

**I. Choose the correct answer from the given options and write the answer with the appropriate letter. (1 mark each):**

1. The pH value range of an acid:

- A) 0-7
- B) 2 -12
- C) 7-14
- D) 12-14

**Answer:**A)0- 7

2. As the number of hydrogen ions increases in a solution:

- A) It becomes neutral
- B) The basicity increases
- C) The acidity increases
- D) The acidity decreases

**Answer:**C)The acidity increases

3. The gas released when dilute sulfuric acid is added to a metal:

- A) Oxygen
- B) Nitrogen
- C) Hydrogen
- D) Carbon

**Answer:**C)Hydrogen

**4. The products formed when an acid is mixed with a metal:**

- A) Metal oxide & water
- B) Metal oxide & hydrogen gas
- C) Salt & water
- D) Salt & hydrogen gas

**Answer:**D)Salt & hydrogen gas

**5. The molecular formula of sodium zincate:**

- A) NaZnO

B) NaZn

C) NaZnO<sub>2</sub>

D) NaZnO<sub>3</sub>

**Answer:**D)NaZnO<sub>3</sub>

**6. The molecular formula of lime water:**

A) CaO

B) Ca(OH)<sub>2</sub>

C) CaCO<sub>3</sub>

D) CO<sub>2</sub>

**Answer:**B)Ca(OH)<sub>2</sub>

**7. The chemical equation that represents neutralization:**

A) H<sub>2</sub>+ O<sub>2</sub>→H<sub>2</sub>O

B) C+O<sub>2</sub>→CO<sub>2</sub>

C) HCl+NaOH→ NaCl+H<sub>2</sub>O

D) Mg+O<sub>2</sub>→MgO

**Answer:**C)HCl+NaOH→ NaCl+H<sub>2</sub>O

**8. The reason for the blue-green color when hydrochloric acid is added to copper oxide solution:**

A) Copper oxide

B) Copper hydroxide

C) Water

D) Copper chloride

**Answer:**D)Copper chloride

**9. The products formed when sodium hydroxide is mixed with water:**

A) Sodium & water

- B) Hydroxide&water
- C) Sodium&hydroxide
- D) Sodium&oxide

**Answer:**C)Sodium&hydroxide

**10. Acidshouldbeaddedtewaterandnotwatertoacidbecause:**

- A) Itisanexothermicreactionandcancauseanexplosion.
- B) Itisanendothermicreactionandcancauseanexplosion.
- C) Itwillnotmix.
- D) Itmixesveryslowly.

**Answer:**A)Itisanexothermicreactionandcancauseanexplosion.

**11. Theacidproducedinthestomach:**

- A) Sulfuricacid
- B) Hydrochloricacid
- C) Nitricacid
- D) Methanoic acid

**Answer:**B)Hydrochloricacid

**12. Amildbaseusedasanantacidtoneutralizeexcessacidityinthestomach:**

- A) Magnesiumhydroxide
- B) Sodiumhydroxide
- C) Potassiumhydroxide
- D) Calciumhydroxide

**Answer:**A)Magnesiumhydroxide

**13. Theacidfoundinnettleplants:**

- A) Ethanoicacid



- B) Methanoic acid
  - C) Propanoic acid
  - D) Tartaric acid
- Answer:** B) Methanoic acid

**14. The reason for using toothpaste on teeth:**

- A) It is an acid
  - B) It sweetens the mouth
  - C) It is a base
  - D) For taste
- Answer:** C) It is a base

**15. Brine solution refers to:**

- A) Sodium hydroxide solution
  - B) Sodium chloride solution
  - C) Potassium hydroxide solution
  - D) Potassium chloride solution
- Answer:** B) Sodium chloride solution

**16. The product of the Chlor-Alkali process used in the production of artificial butter:**

- A) Chlorine gas
  - B) Sodium hydroxide
  - C) Sodium chloride
  - D) Hydrogen gas
- Answer:** D) Hydrogen gas

**17. The salt used as a raw material in glass manufacturing:**

- A) Chalk powder

- B) Washingsoda
  - C) Bakingsoda
  - D) Tablesalt
- Answer:**B)Washingsoda

**18. ThechemicalnameofPlasterofParis:**

- A) Calciumsulfate
  - B) Calciumsulfatehemihydrate
  - C) Calciumsulfonate
  - D) Potassiumsulfate
- Answer:**B)Calciumsulfatehemihydrate

**19. Thereasonwhyblue-coloredcoppersulfateturnswhitewhenheated:**

- A) Duetogainingwater
  - B) Duetoexcesswater
  - C) Duetolosingwater
  - D) Duetodissolvinginwater
- Answer:**C)Duetolosingwater

**20. Thecauseoftoothdecay:**

- A) Acidproducedinthemouth
  - B) Baseproducedinthe mouth
  - C) Saltproducedinthemouth
  - D) Drinkingwater
- Answer:**A)Acidproducedinthemouth

**II. Answerthefollowingquestionsinonesentence: (1markeach)**

**21. Howcanyoudetectanacidusinglitmuspaper?**

**Answer:**Bluelitmuspaperturnsredinacid,whilereditlitmuspaperdoesnot change color in acid.

**22. Whatisan acid?**

**Answer:**AsolutionwithapHrangefrom0to7iscalledanacid.

**23. Write a chemical equation showing the reaction of a metal with a base. Answer:**  $2\text{NaOH} + \text{Zn} \rightarrow \text{Na}_2\text{ZnO}_2 + \text{H}_2$

**24. What precipitate is formed when air is blown into limewater? Answer:** Calcium carbonate ( $\text{CaCO}_3$ )

**25. Which gas is released when acids react with metal carbonates and metal hydrogen carbonates? Answer:** Carbon dioxide ( $\text{CO}_2$ )

**26. What are the products formed when carbonates and hydrogen carbonates react with acids? Answer:** Salt, carbon dioxide, and water.

**27. Why are metal oxides called basic oxides? Answer:** Metal oxides are called basic oxides because they react with acids to form salt and water, similar to a base.

**28. Why are non-metal oxides called acidic oxides? Answer:** Non-metal oxides are called acidic oxides because they react with bases to form salt and water, similar to an acid.

**29. What are alkalis? Answer:** Alkalis are bases that dissolve in water.

**30. Which ion is produced when an acid dissolves in water? Answer:** Hydronium ion ( $\text{H}_3\text{O}^+$ )

**31. What happens when the number of hydroxide ions in a solution increases? Answer:** The basicity (or alkalinity) of the solution increases, or the pH value increases.

**32. What is the pH value of acid rain? Answer:** Less than 5.5

**33. What is the chemical name of bleaching powder? Answer:** Calcium oxychloride

**34. Write the chemical equation for the preparation of bleaching powder. Answer:**  $\text{Ca}(\text{OH})_2 + \text{Cl}_2 \rightarrow \text{CaOCl}_2 + \text{H}_2\text{O}$

**35. What is baking powder? Answer:** Baking powder is a mixture of baking soda and tartaric acid used in cooking.

**III. Answer the following questions in two sentences: (2 mark each)**

**39. Write two differences between acids and bases.**

<b>Answer:</b>	<b>Acids</b>	<b>Bases</b>
1.	Have a sour taste	Have a bitter taste
2.	pH range from 0 to 7	pH range from 7 to 14
3.	Have excess hydrogen ions	Have excess hydroxide ions
4.	Blue litmus turns red	Red litmus turns blue
5.	Example: HCl	Example: NaOH

**40. What are the uses of chlorine gas produced in the chlor-alkali process?**

**Answer:** Water purification, PVC production, disinfectants, CFC production, insecticide manufacturing.

**41. Write two uses of sodium hydroxide.**

**Answer:** Removing stubborn stains from metals, manufacturing soaps and detergents, paper and synthetic fiber production.

**IV. Answer the following questions: (3/4 mark each)**

**42. List three uses of washing soda.**

**Answer:**

- A) Used in glass, soap, and paper industries
- B) Used in the production of sodium compounds like borax
- C) Used as a household cleaner
- D) Used in the removal of permanent hardness of water

**43. List three uses of bleaching powder.**

**Answer:**

- A) Used to bleach cotton and linen in fabric industries
- B) Used to bleach wood pulp in paper industries
- C) Used to bleach laundry in laundries
- D) Used as an oxidizing agent in chemical industries to purify drinking water from germs.

44. Write the chemical equation for the preparation of gypsum and list two uses.

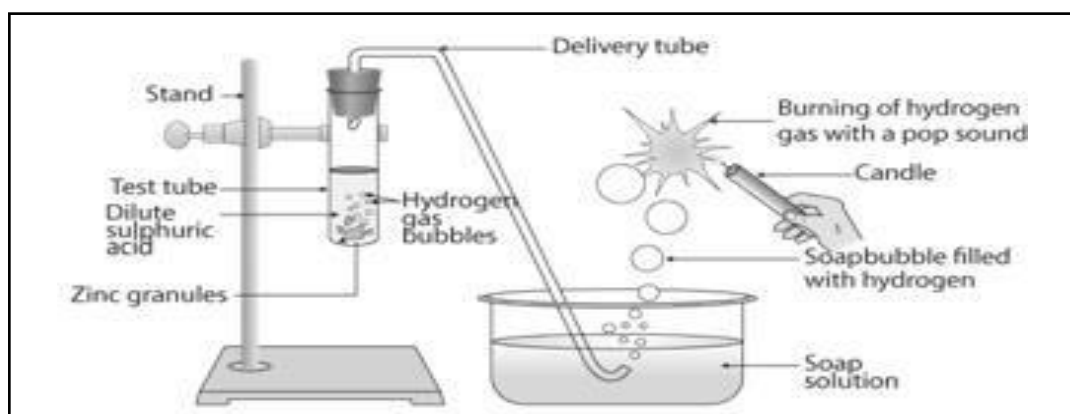
**Answer:**

**Chemical Equation:**  $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O} + \frac{1}{2}\text{H}_2\text{O} \rightarrow \text{CaSO}_4 \cdot 2\text{H}_2\text{O}$

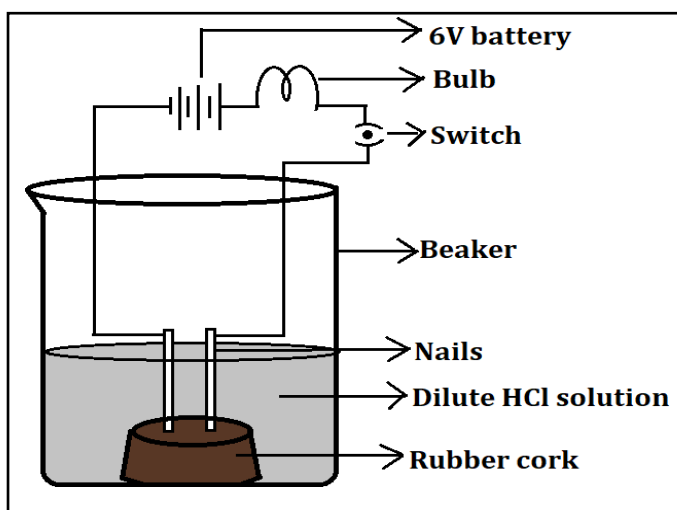
**Uses:**

- Manufacture of plaster
- Production of decorative items
- Creating smooth surfaces

45. Draw the diagram for the reaction of sulfuric acid with iron filings and the test for hydrogen gas through combustion. Label the parts.



46. Draw the diagram for the electrolysis of an acidic solution and label the parts



### Chapter-wise Questions: Chapter-3: Metals and NonMetals

I. Choose the correct answer for each of the following questions, selecting from the four options given. (1 mark per question)

1. The metal with the highest density

- a) Gold
- b) Copper
- c) Iron
- d) Lead

Answer: a) Gold

2. Metals with the highest ductility

- a) Gold
- b) Copper
- c) Silver
- d) Both a and c

Answer: d) Both a and c

3. The best conductor of heat

- a) Gold

- b) Copper
- c) Silver
- d) Bothbandc

Answer:d)Bothbandc

4. Thepoorconductorsofheat

- a) Lead
- b) Rubber
- c) Silver
- d) Bothaandb

Answer:d)Bothaandb

5. Theinsulationonelectricalwiresismadeofplasticorrubberbecause

- a) Theylookattractive
- b) Theyhaveinsulatingproperties
- c) Theyarereadilyavailable
- d) Theyaremoreexpensive

Answer:b)Theyhaveinsulatingproperties

6. Thesubstancethatglows

- a) Iodine
- b) Dye
- c) Sulfur
- d) Carbon

Answer:a)Iodine

7. Thergroupofalkalimetals

- a) Sodium,Iron,Lithium

- b) Sodium, Potassium, Iron
- c) Sodium, Dye, Potassium
- d) Sodium, Lithium, Potassium

Answer: d) Sodium, Lithium, Potassium

8. A form of carbon with a hard, crystalline structure

- a) Graphite
- b) Diamond
- c) Fullerene
- d) Charcoal

Answer: b) Diamond

9. Metals with the lowest melting points

- a) Gallium and Cesium
- b) Sodium and Potassium
- c) Sodium and Gallium
- d) Potassium and Cesium

Answer: a) Gallium and Cesium

10. Products formed when metals dissolve in water

Answer: b) Acidic oxides

11. Amphoteric oxide is

- a) Potassium oxide
- b) Sodium oxide



c) Magnesiumoxide

d) Aluminumoxide

Answer:d)Aluminumoxide

12. Productsformedwhenmetalsreactwithoxygen

a) Alloyoxides

b) Metalhydroxides

c) Metaloxides

d) Alloy hydroxides

Answer:c)Metaloxides

13. Metalsthatreactwithair

a) Iron b) Aluminumc) Zinc d)Alloftheabove

Answer: d) All of the above

14. Thereasoncalciumfloatsonwateristhatcalciumreactswithwaterto produce

a) Hydrogen

b) Oxygen

c) Chlorine

d) Heat

Answer:a)Hydrogen

15. Metalsthatreactwithcoldwatertoreleaseheat

a) SodiumandIron

b) SodiumandPotassium

c) IronandCopper

d) CopperandZinc

Answer:b)SodiumandPotassium

16. Metalthatreactswithhotwater

a) Iron

b) Gold

c) Copper

d) Magnesium  
Answer:d)Magnesium

17. The most powerful oxidizing agent

a) HNO<sub>3</sub>  
b) H<sub>2</sub>SO<sub>4</sub>  
c) CO  
d) KMnO<sub>4</sub>  
Answer:a)HNO<sub>3</sub>

18. The most reactive metals

a) Sodium  
b) Titanium  
c) Potassium  
d) Both a and c  
Answer:d)Both a and c

19. The least reactive metals chemically

a) Lead  
b) Gold  
c) Silver  
d) All of the above  
Answer:d)All of the above

20. The equation indicating the reaction between metals and acids

a)  $\text{Na} + \text{Cl}_2 \rightarrow \text{NaCl}$   
b)  $\text{H}_2 + \text{Cl}_2 \rightarrow \text{HCl}$   
c)  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$   
d)  $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$   
Answer:a) $\text{Na} + \text{Cl}_2 \rightarrow \text{NaCl}$

21. Metals found in their free state

a) Gold, Silver, Iron, and Copper  
b) Gold, Silver, Lead, and Copper

c) Gold, Silver, Platinum, and Copper

d) Gold, Silver, Tin, and Copper

Answer: c) Gold, Silver, Platinum, and Copper

22. TheoreofMercuryanditsformula

a) Chalcopyrites-CuSO<sub>4</sub>

b) CopperGlance -CuO

c) Hematite-Fe<sub>2</sub>O<sub>3</sub>

d) Cinnabar-HgS

Answer: d) Cinnabar-HgS

23. Electrolyticsolutionusedinthepurificationofcopper

a) CuSO<sub>4</sub>

b) CuO

c) Cu<sub>2</sub>O

d) CuCl<sub>2</sub>

Answer: a) CuSO<sub>4</sub>

24. Chemicalprocessusedtoweldrailwaytracks

a) Electrolysis

b) Thermitereaction

c) Metallurgy

d) Electroplating

Answer: b) Thermitereaction

II. Answerthefollowingquestionsinonesentence.(1markperquestion)

25. Whatarealkalimetals?

Answer: Alkalimetalsarethosethatreactwithwatertoproducealkalis.

26. Whatisanallotrope?

Answer: Anallotropeisaphenomenonwhereanelementoccursinmore than one form in nature.

27. What is the allotrope of carbon that is a good conductor

Answer Graphite

28. Which metal is in liquid form at room temperature

Answer: Mercury

29. What is a sonorous metal

Answer A metal that produces a sound when struck on a hard surface is called a sonorous metal.

30. What is an amphoteric oxide?

Answer: Amphoteric oxides are metal oxides that react with both acids and bases to produce salts and water.

31. State one property of alkali metals

Answer Alkali metals are very soft and can be cut with a knife.

32. What is anode mud?

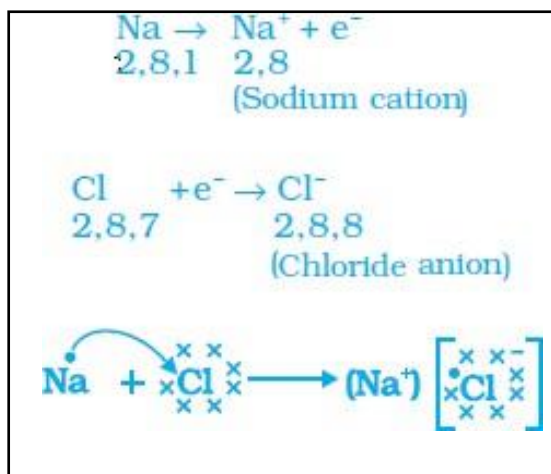
Answer: In electrolysis, the insoluble impurities that settle at the bottom of the anode are called anode mud.

33. What is galvanization?

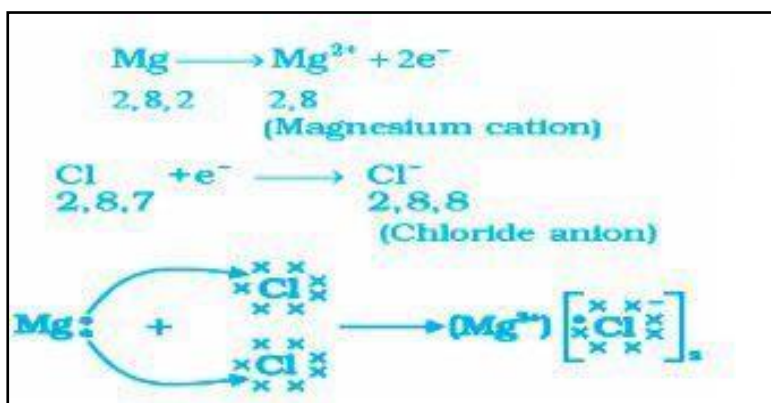
Answer: Galvanization is the process of coating iron and steel with a layer of zinc to protect them from rusting.

III. Answer the following questions in two sentences. (2 marks per question)

34. Write the electron configuration for the formation of sodium chloride.



35. Write the configuration for the formation of magnesium chloride.



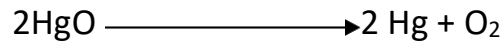
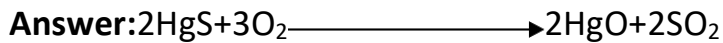
36. What is an alloy? Give one example.

Answer: An alloy is a homogeneous mixture of two or more metals. An example is bronze, which is a mixture of copper and tin.

37. Which component metals are present in solder used for electrical connections? Why is solder used in electrical connections?

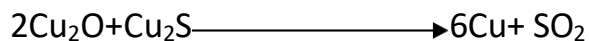
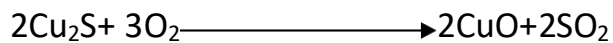
Answer: Solder typically contains lead and tin. It is used in electrical connections because it melts at a low temperature and solidifies quickly, creating a strong, conductive joint.

**38. Write the chemical equations for obtaining copper from cinnabar.**



**39. Write the chemical equations for separating copper from copper sulfide.**

**Answer:**



**40. Write the difference between roasting and calcination.**

**Answer:**

Roasting	Calcination
1. Converting sulfide ores into oxides.	1. Converting carbonate ores into oxides.
2. Performed in the presence of excess air.	2. Performed in the presence of limited air.
3. Example: $\text{ZnS} + 3\text{O}_2 \rightarrow 2\text{ZnO} + 2\text{SO}_2$	3. Example: $\text{ZnCO}_3 \rightarrow \text{ZnO} + \text{CO}_2$

**41 Physical Differences Between Metals and Alloys**

Metals	Alloys
1. They shine or have a luster.	1. Do not shine.

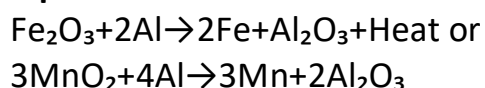
- |                                    |  |
|------------------------------------|--|
| 2. Solid at room temperature.      | 2. Can be in solid, liquid, or gaseous state at room temperature |
| 3. Good conductors of electricity. | 3. Poor conductors of electricity or non-conductors.             |
| 4. Good conductors of heat.        | 4. Poor conductors of heat or insulators.                        |
| 5. Malleable.                      | 5. Not malleable.  |
| 6. Ductile.                        | 6. Not ductile.  |

42. What is the thermite reaction? Give examples

**Thermite Reaction**

**Definition:** The thermite reaction is a high-temperature reaction that involves mixing a more reactive metal with metal oxides to separate the metal from the compound, releasing a significant amount of heat.

**Equation:**



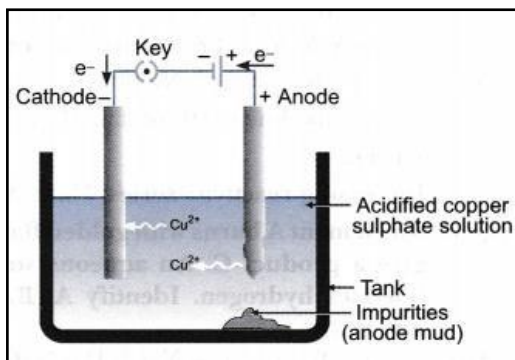
**IV. Answer the following questions. (Each question carries 03/04 marks)**

**43. Write the properties of ionic compounds**

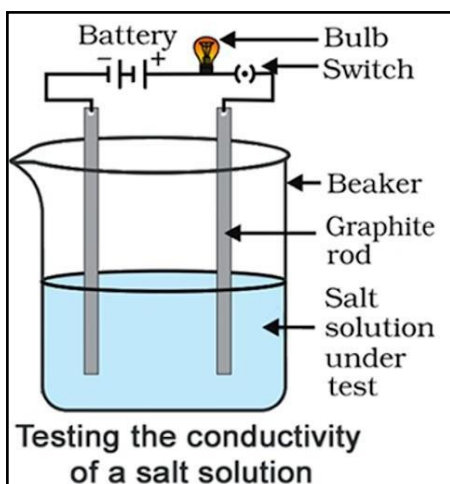
**Properties of Ionic Compounds:**

- They exist as solids and are hard.
- They have high melting and boiling points.
- They dissolve in water but not in organic solvents.
- Their aqueous solutions conduct electricity, but they do not conduct electricity in the solid state.

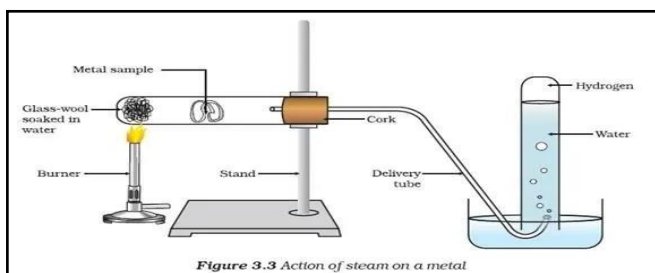
**44. Diagram of Electrolysis of Copper(II) Sulfate Solution:**



45. Diagram of Testing the Conductivity of a Salt Solution:



46. Diagram of Metal Reacting with steam





Chapter-wise Questions: Chapter 4 - Carbon and Its Compounds

I. For each of the following questions, four options are provided. Choose the correct answer and write it with the sequence letter. (1 mark per question)

1. Valency of Carbon

- a) 1
- b) 2
- c) 3
- d) 4

**Answer: d) 4**

2. Aromatic Carbon Compound

- a) Carbon atoms are bonded by triple bonds
- b) Carbon atoms are bonded by double bonds
- c) Carbon atoms are bonded by single bonds
- d) There are no bonds between carbon atoms

**Answer: c) Carbon atoms are bonded by single bonds**

3. Maximum number of bonds a single Carbon atom can form with Hydrogen atoms

- a) 1
- b) 2
- c) 3
- d) 4

**Answer: d) 4**

4. Number of Carbon atoms in Ethene

- a) 1
- b) 2
- c) 3
- d) 4

**Answer: b) 2**

5. First member of Alkanes

- a) Methane
- b) Ethane
- c) Propane
- d) Butane

**Answer: a) Methane**

6. **First member of Alkenes**

- a) Methane
- b) Ethane
- c) Ethene
- d) Propene

**Answer: c) Ethene**

7. **First member of Alkynes**

- a) Propyne
- b) Ethyne
- c) Butyne
- d) Pentyne

**Answer: b) Ethyne**

8. **Difference between any two consecutive Alkanes/Alkenes/Alkynes**

- a) CH<sub>2</sub>
- b) CH
- c) CH<sub>3</sub>
- d) CH<sub>4</sub>

**Answer: a) CH<sub>2</sub>**

9. **Suffix of Aromatic Carbon Compounds**

- a) -ene
- b) -ine
- c) -ane
- d) -ol

**Answer: c) -ane**

10. **Suffix of Non-Aromatic Carbon Compounds**

- a) -ene
- b) -ine
- c) -ane
- d) Both a and b

**Answer: d) Both a and b**

11. **Suffix for carbon compound having the Aldehyde functional group**

- a) -ol
- b) -one
- c) -al
- d) -oic acid

**Answer: c) -al**

12. Catalyst used in the hydrogenation of oils

- a) Nickel
- b) Chromium
- c) Cobalt
- d) Lead

Answer: a) Nickel

13. Oxidizing agents used to convert ethanol to acetic acid

- a)  $\text{KMnO}_4$
- b)  $\text{K}_2\text{Cr}_2\text{O}_7$
- c) Either a or c
- d) Nickel

Answer: c) Either a or c

II. Answer the following questions in one sentence each (1 mark per question)

14. What is the simplest carbon compound?

Answer: Methane

15. Functional group present in acetic acid

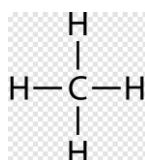
Answer: Carboxylic group

16. What is glacial acetic acid?

Answer: Acetic acid that solidifies in cold weather is called glacial acetic acid.

17. Write the molecular and structural formula of methane.

Answer:  $\text{CH}_4$



18. How many covalent bonds are there in ethane?

Answer: 7

19. General formula for alkanes

Answer:  $C_nH_{2n+2}$

20. General formulas for alkenes and alkynes

Answer:  $C_nH_{2n}$  &  $C_nH_{2n-2}$

21. What is saponification?

Answer: Saponification is the process of producing organic salts by reacting fatty acids with alkalis.

22. What is a micelle?

Answer: A micelle is the structure formed when the ionic ends of soap molecules interact with water while the carbon chains interact with oil.

23. What is esterification?

Answer: Esterification is the process of forming an ester by reacting an alcohol with a carboxylic acid.

23. What is esterification?

**Answer:** Esterification is the process of forming esters from the reaction between an acid and an alcohol.

24. What is a substitution reaction?

**Answer:** A substitution reaction is a process where atoms or groups of atoms in a molecule are replaced by different atoms or groups of atoms.

25. What are the possible products when carbon compounds are burned?

**Answer:** Carbon dioxide, water, heat, and light.

**III. Answer the following questions in two sentences each (2 marks per question)**

26. Write the differences between saturated and unsaturated carbon compounds.

**Answer:**

**Saturated Carbon Compounds**

1. Single bonds between carbon atoms
2. First member is methane
3. General formula:  $C_nH_{2n+2}$

**Unsaturated Carbon Compounds**

1. Double or triple bonds between carbon atoms
2. First members are ethene/ethyne
3. General formulas:  $C_nH_{2n}$  or  $C_nH_{2n-2}$

27. What are isomers? Who is the first member of isomers?

**Answer:** Isomers are carbon compounds with the same molecular formula but different structural formulas. The first member of isomers is iso-butane.

28. What are haloalkanes? Give an example.

**Answer:** Haloalkanes are carbon compounds that contain halogen groups.  
Example: Chloro-methane.

29. What are functional groups?

**Answer:** Functional groups are groups of different atoms that replace hydrogen in carbon compounds and determine the type of chemical reaction the compound undergoes.

30. What are hydrocarbons? How many types are there? Name them.

**Answer:** Hydrocarbons are compounds made of carbon and hydrogen. There are three types: alkanes, alkenes, and alkynes.

# "Biology"

"Chapter:LifeProcesses"

1. Theroleofchlorophyllinphotosynthesisis\_\_\_\_\_

- a) Applyingupwardpressure
- b) Absorbingcarbondioxide
- c) Releasingoxygen
- d) Absorbing light energy.

Answer:d)Absorbinglightenergy.

2. Inphotosynthesis,carbondioxideisconvertedinto\_\_\_\_\_.

- a) Carbohydrates
- b) Proteins
- c) Vitamins
- d) Lipids.

Answer:a)Carbohydrates.

3. Thisnutrientisessentialforautotrophyinplantsandisusedinprotein synthesis. \_\_\_\_\_

- a) Oxygen
- b) Nitrogen
- c) Carbon
- d) Phosphorus.

Answerb)Nitrogen.

4. The saliva in the mouth breaks down starch into sugar with the help of this enzyme \_\_\_\_\_

- a) Pepsin
- b) Amylase
- c) Renin
- d) Trypsin

Answer: b) Amylase.

5. The acid secreted by the stomach is \_\_\_\_\_

- a) Nitric acid
- b) Sulfuric acid
- c) Acetic acid
- d) Hydrochloric acid

Answer: d) Hydrochloric acid.

6. Birds excrete nitrogenous wastes in this form \_\_\_\_\_

- a) Carbon dioxide
- b) Urea
- c) Ammonia
- d) Uric acid

Answer: d) Uric acid.

7. The enzyme that secretes mucus to protect the stomach lining from the acid secreted in the stomach is \_\_\_\_\_

- a) Pepsin
- b) Amylase
- c) Renin
- d) Trypsin

Answer: a) Pepsin.

8. Due to these substances in their diet, the small intestine of herbivores is longer than that of carnivores. \_\_\_\_\_

- a) Protein
- b) Carbohydrate
- c) Cellulose
- d) Lipid

Answer: c) Cellulose.

9. The part of the digestive system where carbohydrates, fats, and proteins are fully digested.

- a) Mouth
- b) Small intestine
- c) Large intestine
- d) Stomach

Answer: b) Small intestine.

10. The finger-like structure that helps in the absorption of digested food from the small intestine is

- a) Enzymes
- b) Villi
- c) Appendix
- d) Liver

Answer: b) Villi.

11. The organ that produces urea is \_\_\_\_\_

- a) Pancreas
- b) Excretory system
- c) Lungs
- d) Skin

Answer: b) Excretory system.



12. The exchange of substances between blood and cell takes place through \_\_\_\_\_.

- a) Veins
- b) Arteries
- c) Capillaries
- d) Valves

Answer: c) Capillaries

13. The component of blood that helps in transport is \_\_\_\_\_.

- a) Plasma
- b) Red blood cells
- c) White blood cells
- d) Platelets

Answer: a) Plasma

14. The role of stomata in transportation in plants is \_\_\_\_\_.

- a) Creating upward pressure
- b) Absorbing carbon dioxide
- c) Releasing oxygen
- d) Absorbing light energy

Answer: b) Absorbing carbon dioxide

15. The acid released in muscles due to the breakdown of glucose in the absence of oxygen is \_\_\_\_\_.

- a) Acetic acid
- b) Citric acid
- c) Lactic acid
- d) Tartaric acid

Answer: c) Lactic acid

16. The respiratory pigment in humans is \_\_\_\_\_

- a) diaphragm
- b) Air sac
- c) Alveoli
- d) Hemoglobin

Answer: d) Hemoglobin

17. This component of blood transports food, carbon dioxide, and nitrogenous wastes.

- a) Plasma
- b) Red blood cells
- c) White blood cells
- d) Platelets

Answer: a) Plasma

18. Oxygen-rich blood enters this chamber of the heart. \_\_\_\_\_

- a) Left atrium
- b) Right atrium
- c) Left ventricle
- d) Right ventricle

Answer: a) Left atrium

19. Oxygen-depleted blood enters this chamber of the heart. \_\_\_\_\_

- a) Left atrium
- b) Right atrium
- c) Left ventricle
- d) Right ventricle

Answer: b) Right atrium

20. The part of the heart that prevents the backflow of blood when the atria and ventricles contract is \_\_\_\_\_

- a) Septum
- b) Arteries
- c) Veins
- d) Valves

Answer: d) Valves

21. The wall that separates the left ventricle and right ventricle of the heart is \_\_\_\_\_

- a) Septum
- b) Arteries
- c) Veins
- d) Valves

Answer: a) Septum

22. In this animal, blood passes through the heart only once. \_\_\_\_\_

- a) Reptiles
- b) Fish
- c) Mammals
- d) Amphibians

Answer: b) Fish

23. The blood component responsible for clotting is \_\_\_\_\_

- a) Plasma
- b) Red blood cells
- c) White blood cells
- d) Platelets

Answer: d) Platelets

24. The fundamental filtration unit of the kidney is \_\_\_\_\_

- a) Ureter
- b) Kidney
- c) Nephron
- d) Urethra

Answer: c) Nephron

25. Name the enzymes secreted by pancreatic juice.

Answer: Trypsin, which digests proteins, and Lipase, which breaks down fats.

26. Name the final products formed by proteins, carbohydrates, and fats.

Answer:

- Proteins: Amino acids
- Carbohydrates: Glucose
- Fats: Glycerol and fatty acids

27. What is pyruvate?

Answer: The process of breaking down the six-carbon molecule of glucose into two three-carbon molecules.

28. What are the different processes of glucose breakdown?

Answer: The different processes of glucose breakdown are:

- 1) Aerobic respiration in the presence of oxygen
- 2) Anaerobic respiration in the absence of oxygen
- 3) Respiration in low oxygen conditions

29. What substances are released when glucose is broken down in the absence of oxygen?

Answer: When glucose is broken down in the absence of oxygen, ethanol, carbon dioxide, and a small amount of energy are produced.

30. Why do aquatic organisms have a high rate of respiration?

Answer: Aquatic organisms have a higher rate of respiration because the amount of oxygen available in water is low, so they breathe rapidly to meet their oxygen requirements.

31. During respiration, the chest cavity expands and air is drawn into the lungs. Which body parts are responsible for this effect?

Answer: During respiration, the expansion of the chest cavity, allowing air to be drawn into the lungs, is caused by the ribs and the diaphragm.

32. How is carbon dioxide transported in the body?

Answer: Carbon dioxide dissolves in water and combines with the blood, allowing it to be transported.

33. How is oxygen transported in the body?

Answer: Oxygen is transported by binding to hemoglobin in the blood.

34. Birds and mammals have four chambers in their hearts. What is the significance of this?

Answer: The body requires a high amount of energy to maintain its complex biological functions. This is only possible when oxygen-rich blood and oxygen-poor blood are separated. Therefore, they have four chambers in their hearts.

35. What is double circulation?

Answer: The process where oxygen-rich and oxygen-poor blood pass through the heart twice, which is referred to as double circulation.

36. Why does a person's blood pressure decrease when injured?

Answer: The blood pressure decreases due to blood loss from the injury, which leads to a reduction in the heart's pumping capacity.

37. Name the types of blood vessels.

Answer: Arteries, veins, and capillaries.

38. How is an artery different from a vein, and why?

Answer: Arteries have thick walls and are elastic to withstand high pressure from blood pumped by the heart. In contrast, veins have thinner walls because they carry blood at lower pressure as it returns to the heart.

39. How is lymph formed in the body?

Answer: Lymph is formed when plasma, proteins, and blood cells pass through the walls of capillaries into the intercellular space of tissues.

40. What is the function of lymph?

Answer: Lymph transports digested and absorbed fats from the intestine and returns excess fluid from tissues back into the bloodstream.

41. How are water and minerals absorbed by roots transported to the leaves at the top of the plant?

Answer: The stomata in the leaves release excess water through transpiration, creating an upward pressure that helps in the movement of water towards the leaves.

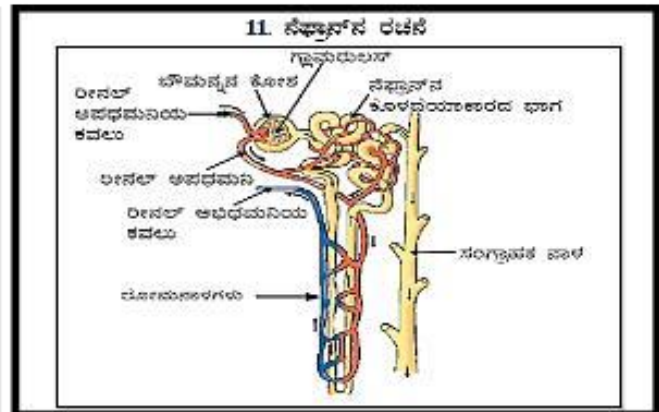
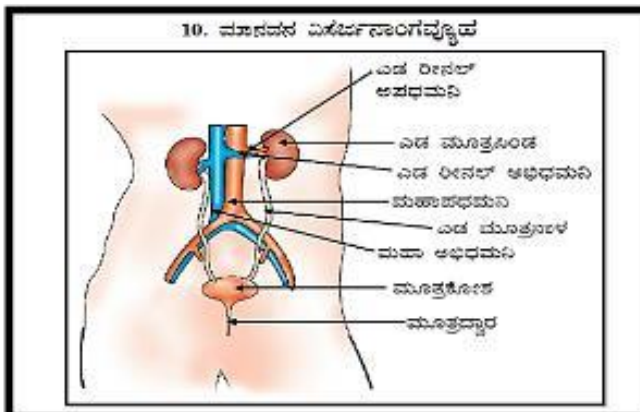
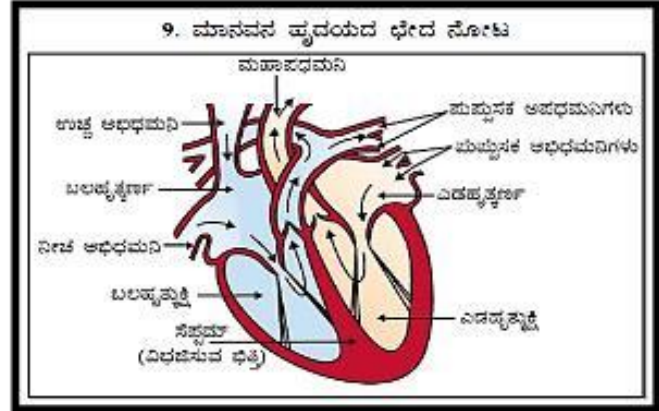
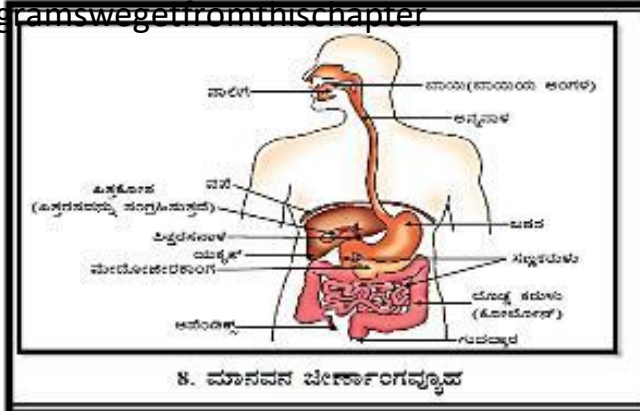
42. What is excretion?

Answer: Excretion is the biological process of removing nitrogenous waste products produced during metabolic activities from the body.

43. How does excretion occur in plants?

Answer: Plants release excess water through transpiration. They store waste products in dead cells or tissues, which eventually fall off. Additionally, some waste is excreted in the form of gums and resins.

Diagrams we get from this chapter



**Control and Coordination**

1. In animals, control and coordination are provided by the \_\_\_\_\_ and \_\_\_\_\_ systems.

- a) Bone & muscle
- b) Nerves & muscles
- c) Bone & nerves
- d) Blood & muscles

Answer: b) Nerves & muscles

2. The information sensed by sensory organs is stored in the \_\_\_\_\_ part of the neuron.

- a) Dendrites
- b) Axon
- c) Nerve ending
- d) Cell body

Answer: a) Dendrites

3. The chemicals released at the axon cross the \_\_\_\_\_ and reach the dendrites of another neuron.

- a) Nerve ending
- b) Axon
- c) Synapse
- d) Cell body

Answer: c) Synapse

4. Reflex action occurs in the \_\_\_\_\_.

- a) Brain
- b) Spinal cord
- c) Neuron
- d) Cranial nerve

Answer: b) Spinal cord

5. The part of the brain that controls activities like riding a bicycle or writing neatly with a pen is the \_\_\_\_\_.

- a) Cerebrum
- b) Cerebellum
- c) Medulla
- d) Pons

Answer: b) Cerebellum



6. When you see eat a marind, your mouth waters. This part of the brain is responsible for it:

- a) Cerebrum
- b) Cerebellum
- c) Medulla
- d) Pons

Answer: c) Medulla

7. The bony structure that protects the spinal cord is the:

- a) Skull
- b) Rib cage
- c) Skeleton
- d) Vertebral column

Answer: d) Vertebral column

8. The movement of pollen toward the ovary in plants is due to:

- a) Phototropism
- b) Geotropism
- c) Hydrotropism
- d) Chemotropism

Answer: d) Chemotropism

9. The hormone that inhibits plant growth is:

- a) Auxin
- b) Cytokinin
- c) Abscisic acid
- d) Gibberellin

Answer: c) Abscisic acid

10. The hormone responsible for phototropism in plants is:

- a) Auxin
- b) Cytokinin
- c) Abscisic acid
- d) Gibberellin

Answer: a) Auxin

11. The information sensed by receptors in the body is transmitted through:

- a) Electrical impulses
- b) Blood
- c) Lymph
- d) Tubes

Answer: a) Electrical impulses

12. The hormone deficiency that causes diabetes is:

- a) Adrenaline
- b) Insulin
- c) Glucagon
- d) Thyroxine

Answer: b) Insulin

13. This hormone is known as the "emergency hormone":

- a) Adrenaline
- b) Insulin
- c) Glucagon
- d) Thyroxine

Answer: a) Adrenaline

14. The deficiency of this hormone causes the Goiter disease:

- a) Adrenaline
- b) Insulin
- c) Glucagon
- d) Thyroxine

Answer: d) Thyroxine

15. The presence of this substance in food keeps us away from Goiter disease:

- a) Carbohydrate
- b) Sodium
- c) Iodine
- d) Iron

Answer: c) Iodine

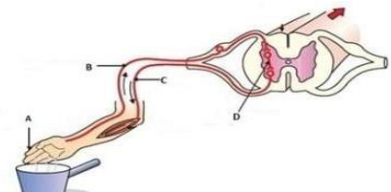
16. The injection doctors give to diabetics is:

- a) Insulin
- b) Tetanus Toxoid (T.T.)
- c) UUU
- d) Estrogen

Answer: a) Insulin

17. What is a reflex action?

Answer: A reflex action is an immediate response to a stimulus.



18. What is a reflex arc?

Answer: From the time an action occurs until a response is given, the neural pathway followed is called a reflex arc.

19. How do muscle cells create movement?

Answer: Muscle cells change their shape to create movement.

20. How do living cells change their shape?

Answer: Muscle cells contain special proteins that respond to electrical impulses from nerves, altering both the shape and structure of the cells.

21. Explain the difference between voluntary and involuntary muscles.

Answer:

Voluntary Muscles	Involuntary Muscles
Muscles that operate under our control	Muscles that function without our conscious control
Found in the arms and legs	Found in the respiratory tract, oesophagus, and blood vessels

22. Why does the "Touch-Me-Not" plant fold its leaves when touched?

Answer: The plant's cells change their water content, altering their shape. This causes a reduction in the water supply to the leaves, leading to the folding of the plant's leaves.

23. How is phototropism beneficial to plants?

Answer: In two types of movement, shoots respond by bending toward the light, while roots bend away from the light. Leaves need sunlight to produce food, and roots are required to absorb water and minerals.

24. Name the plant hormones and explain their functions.

Answer: Auxin, Cytokinin, Gibberellin, and Abscisic acid are plant hormones.

Plant Hormone	Function
Auxin	Stimulates growth
Cytokinin	Promotes cell division
Gibberellin	Aids in stem growth
Abscisic Acid	Inhibits plant growth

25. What is phototropism?

Answer: It is the behaviour of a plant bending and growing towards the light.

26. What is geotropism?

Answer: Geotropism is the downward growth of plant roots due to the effect of gravitation.

27. What is the function of receptors in the human body? Name their types.

Answer: Receptors perform the function of perceiving various external stimuli. They are also called sensory organs. Their types are:

- 1) Olfactory receptors in the nose detect smell.
- 2) Visual receptors in the eyes detect sight.
- 3) Auditory receptors in the ears detect sound.
- 4) Gustatory receptors on the tongue detect taste.
- 5) Touch receptors in the skin detect touch.



28. What happens if receptors stop functioning properly?

Answer: Receptor cells are located in sensory organs. They perceive changes in the environment and trigger appropriate responses. If they stop functioning, actions and reactions do not occur. For example, when we touch something hot, we immediately pull our hand back to avoid injury. If this response fails, the hand may get burned.

29. Name the part that helps in transmitting nerve impulses in a neuron. The

part that helps transmit nerve impulses in a neuron:

- a) Towards the cell: Dendrite
- b) Away from the cell: Axon

30. Briefly represent the transmission of nerve impulses in the body using a flow chart.

Answer:

Information → Reaches the dendrites of the neuron → Electrical signals → Cell body → Axon → Axon terminal → Information is shared at the synapse → Finally, impulses reach the target organ from the neuron, initiating a response.

31. A student, due to an accident, is experiencing the following issues:

- 1) Unable to stand properly.
- 2) Lost the ability to sense smell.
- 3) Unable to taste food.

Indicate which part of the brain is affected in each case. Answer:

- 1) Unable to stand properly – The cerebellum is affected, causing a loss of body balance.
- 2) Lost the ability to sense smell – The frontal lobe, which is responsible for the sense of smell, is affected.
- 3) Unable to taste food – The frontal lobe, which is responsible for the sense of taste, is affected.

32. Explain the difference between nervous system control and chemical control.

Answer:

Nervous System Control

Control is achieved through specific organs such as nerves.

The information received by receptors reaches the target organ through

Chemical Control

There are no specific organs; control is achieved through secreted chemicals called hormones.

Information reaches the target organ through hormones.



3. An example of reproduction by regeneration is\_\_\_\_\_.

- a) Leishmania
- b) Spirogyra
- c) Hydra
- d) Planaria

Answer:d)Planaria

4. An example of reproduction by budding is\_\_\_\_\_.

- a) Leishmania
- b) Spirogyra
- c) Hydra
- d) Planaria

Answer:c)Hydra

5. An example of reproduction by vegetative propagation is\_\_\_\_\_.

- a) Leishmania
- b) Spirogyra
- c) Sugarcane
- d) Planaria

Answer:c)Sugarcane

6. An example of reproduction through spore formation is\_\_\_\_\_.

- a) Leishmania
- b) Spirogyra
- c) Sugarcane
- d) Rhizopus

Answer:d)Rhizopus



7. The reproductive part of a plant is \_\_\_\_\_.

- a) Root
- b) Stem
- c) Flower
- d) Leaf

Answer: c) Flower

8. The part of the flower that produces pollen is \_\_\_\_\_.

- a) Stigma
- b) Petals
- c) Anther
- d) Style

Answer: c) Anther

9. An example of reproduction by fragmentation is \_\_\_\_\_.

- a) Leishmania
- b) Spirogyra
- c) Sugarcane
- d) Rhizopus

Answer: b) Spirogyra

10. The correct reason why the testes in males are located outside the body is \_\_\_\_\_.

- a) To facilitate the smooth movement of sperm
- b) So that sperm can move smoothly due to body heat

c) The production of sperm requires a temperature lower than body temperature

d) The production of sperm requires a temperature higher than body temperature

Answer: c) The production of sperm requires a temperature lower than body temperature

11. In females, the ovary produces \_\_\_\_\_ egg every month.

a) 10

b) 1

c) 20

d) 15

Answer: b) 1

12. When the foetus is in the womb, it receives nutrients from the mother through ----.

a) Amniotic fluid

b) Placenta

c) Umbilical cord

d) Blood vessels

Answer: c) Umbilical cord

13. The virus that causes sexually transmitted disease is \_\_\_\_\_

a) HIV

b) HPV

c) Syphilis

d) Gonorrhoea

Answer: a) HIV

14. The hormonal contraceptive method used to prevent pregnancy is \_\_\_\_\_

- a) Condom
- b) Birth control pills
- c) Surgery
- d) Copper-T

Answer: b) Birth control pills

15. Which of the following is not a part of female reproduction in humans?  
\_\_\_\_\_

- a) Ovaries
- b) Uterus
- c) Vas deferens
- d) Fallopian tubes

Answer: c) Vas deferens

16. The pollen sac is associated with \_\_\_\_\_

- a) Petals
- b) Ovules
- c) Stigma
- d) Pollen grains

Answer: d) Pollen grains

17. The part that produces sperm is \_\_\_\_\_

- a) Vas deferens
- b) Testis
- c) Fallopian tube
- d) Ovaries

Answer: b) Testis

18. The organisms that cause cholera and malaria reproduce by the method of \_\_\_\_\_.

- a) Binary fission
- b) Regeneration
- c) Fragmentation
- d) Budding

Answer: a) Binary fission

19. What is the function of DNA?

Answer: DNA is the source of information for synthesizing proteins. It is responsible for the inheritance of traits from parents to offspring.

20. What is the importance of diversity in living organisms?

Answer: Organisms can go extinct due to changes in the environment. If there is diversity among some organisms in a community, their chances of survival increase.

21. Explain the methods of reproduction with an example for each.

Answer: The methods of reproduction are:

- Binary fission: Amoeba, Leishmania, Plasmodium
- Fragmentation: Spirogyra
- Regeneration: Hydra, Planaria
- Budding: Hydra
- Vegetative reproduction: Sugarcane, Rose, Cabbage, Banana, Bryophyllum
- Spore formation: Rhizopus, Hyphae

22. What is zygote

Answer: Zygote is the stage when the male and female gametes fuse and the egg matures.

23. Name the parts involved in the reproduction of a plant's flower.

Answer: The parts involved in the reproduction of a plant's flower are the anther and stigma.

24. Name the male and female reproductive parts in a plant's flower.

Answer: The male reproductive part in a plant's flower is the anther, and the female reproductive part is the stigma.

25. What is self-pollination?

Answer: Self-pollination occurs when the pollen from a flower is transferred to the stigma of the same flower.

26. What is cross-pollination?

Answer: Cross-pollination occurs when the pollen from one flower is transferred to the stigma of another flower.

27. Explain the difference between self-pollination and cross-pollination.

Answer:

- Self-pollination: This occurs when the pollen from a flower is transferred to the stigma of the same flower.

- Cross-pollination: This occurs when the pollen from one flower is transferred to the stigma of another flower.

28. How does the pollen reach the ovary after contacting the stigma?

Answer: After the pollen contacts the stigma, it reaches the ovary through the style.

29. How is an embryo formed?

Answer: The zygote undergoes several divisions to form the embryo.

30. How is a seed formed?

Answer: The fertilized ovule develops a hard outer layer around itself, and then it gradually grows into a seed.

31. What are the common changes observed in females during puberty?

Answer: The voice becomes softer, the size of the breasts begins to increase, the menstrual cycle starts, and hair growth is observed around the armpits and genitalia.

32. What are the common changes observed in males during puberty?

Answer: The voice breaks, facial hair begins to grow, and growth becomes more pronounced than in females. Hair growth is also seen around the armpits and genitalia.

33. What is puberty?

Answer: Puberty is the period of adolescence, typically between the ages of 13 to 19.

34. What hormone is responsible for physical changes in males during puberty?

Answer: The hormone responsible for physical changes in males during puberty is testosterone.

35. What glands are found along the path of the vas deferens?

Answer: The glands found along the path of the vas deferens are the prostate and seminal vesicles.

36. What happens to an ovum released in females if it is not fertilized?

Answer: If the ovum released in females is not fertilized, it can survive for up to one day. During this time, the uterus prepares to receive the ovum through a spongy structure. However, if the ovum is not fertilized, the lining of the uterus gradually breaks down, leading to the release of blood and tissue through the vagina. This process occurs every month when an ovum is released, so it is called the menstrual cycle.

37. What is the purpose of advising purity until marriage?

Answer: A woman must be physically and mentally capable of handling the important responsibility of pregnancy. Moreover, multiple partners in sexual relationships can lead to several serious diseases, including AIDS, gonorrhoea, and syphilis.

38. What are the methods available to prevent pregnancy?

Answer: There are various methods designed to prevent pregnancy, including the use of condoms, hormonal contraceptive pills that alter hormonal balance, and the insertion of devices like IUDs or Copper-T into the uterus. Pregnancy can also be prevented through surgical methods.

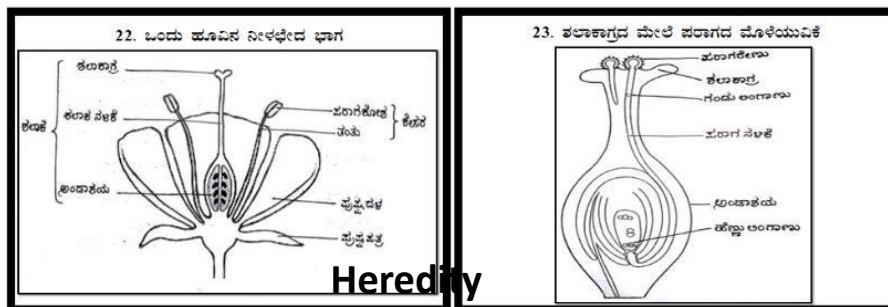
39. How does the embryo receive nourishment inside the mother's womb?

Answer: The embryo receives nourishment through the placenta in the mother's womb.

40. If a woman is using a Copper-T, does it help protect her from sexually transmitted diseases?

Answer: The Copper-T prevents sperm from reaching the egg, but it does not protect against diseases transmitted by bacteria and viruses. If these pathogens come into contact with bodily fluids, they can multiply and spread diseases.

"Images included in this chapter.



1. Individuals with a complete pair of sex chromosomes are \_\_\_\_\_.

- a) Female only
- b) Male only
- c) Both male and female

Dependent on other factors

Answer: a) Female only

2. On which plant did Mendel conduct his study of the laws of inheritance?

- a) Rose
- b) Pea plant
- c) Tomato
- d) Potato

Answer: b) Pea plant

3. In monohybrid crosses, the plants Mendel obtained in the F<sub>1</sub> generation were \_\_\_\_\_.

- a) All dwarf b) All tall c) Some tall only d) Some dwarf only

Answer: b) All tall

4. The ratio of plants obtained in the F<sub>2</sub> generation in a dihybrid cross is \_\_\_\_\_.

- a) 9:3:1:3 b) 9:3:3:1 c) 1:3:3:9 d) 3:1

Answer: b) 9:3:3:1

5. The sex chromosome in females is \_\_\_\_\_.

- a) XX b) XY c) YY d) XO

Answer: a) XX

6. The sex chromosome in males is \_\_\_\_\_.

- a) XX b) XY c) YY d) XO

Answer: b) XY

7. The sex chromosome that determines the birth of a male child is \_\_\_\_\_.

- a) X b) Y c) XX d) YY

Answer: b) Y

8. What is genetics?

Answer: Genetics refers to the scientific study and research related to the reproduction of animals and plants.



9. How is the sex of a child determined in humans?

In humans, there are sex chromosomes present, which are X in females and XY in males. In a fertilization context, either the X or Y chromosome from the male combines with the X chromosome from the female. If the combination is XY, a male child is born; if it is XX, a female child is born. Thus, the sex of the child is determined.

10. What is a monohybrid cross?

Answer: A monohybrid cross refers to the genetic cross between parents that differ in only one trait.

11. What is a dihybrid cross?

Answer: A dihybrid cross refers to the offspring resulting from the crossing of plants that have two different traits.

12. Why are the traits acquired by an organism during its lifetime not hereditary?

Answer: The traits acquired by an organism during its lifetime do not cause changes in the DNA of the gametes, so these traits are not hereditary.

13. Create a checkerboard to show the phenotypic ratio in a monohybrid cross.

Answer:

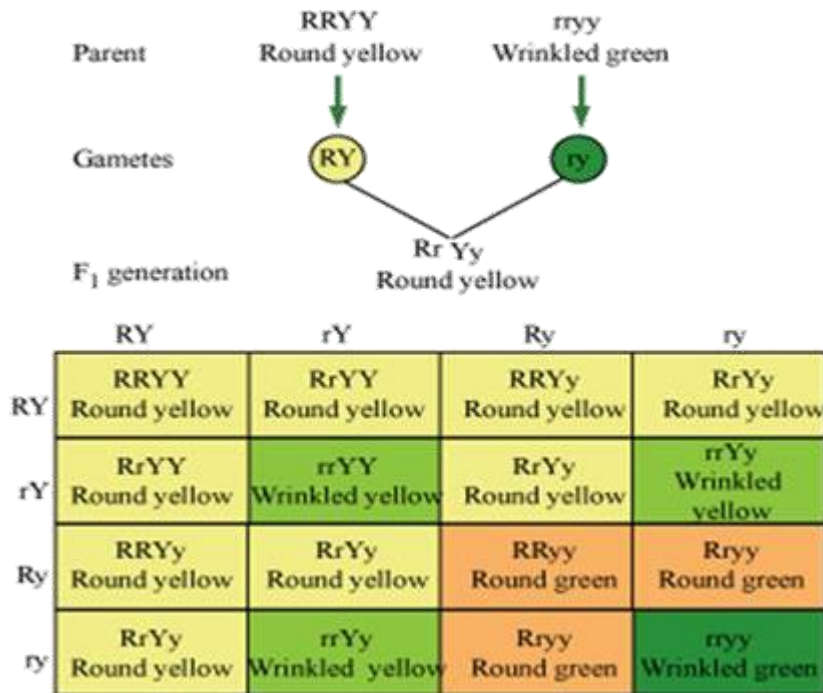
In this F<sub>1</sub> generation, all plants exhibit the tall trait. In

this F<sub>2</sub> generation:

	T	t
T	TT	Tt
T	Tt	tt

(Note: The letters T and t represent the dominant and recessive traits, respectively.)

14. Create a checkerboard to show the phenotypic ratio in a dihybrid cross.



Answer:

The phenotypic ratio is 9:3:3:1.

15. If one trait has a frequency of 10% in an asexual reproduction group and another trait has a frequency of 60% in the same group, which trait could have arisen first?

Answer: The trait that is present in a higher percentage of the population (60%) is likely to have arisen first and spread throughout the population, while the trait with 10% is newer and less widespread.

16. How does the variation among species promote their survival?

Answer: Reproductive processes, especially sexual reproduction, lead to new variations that are similar but subtly different. Thus, different variations emerge within a species. Variations that are suited to the current environment survive through natural selection, while less

suited variations may disappear. As the environment changes, only those variations that are suitable for the new conditions will survive, ensuring the continuity of the species.

17. How do Mendel's experiments show that traits can be dominant or recessive?

Answer: After self-fertilization, in the second generation, 75% of the individuals exhibited the dominant trait, while 25% displayed the recessive trait.

18. How do Mendel's experiments show that traits are inherited independently?

Answer: Mendel crossed two pairs of contrasting traits in pea plants. The first generation (F1) showed only the dominant traits from each pair. In the second generation (F2), the offspring displayed both parental traits as well as new trait combinations not present in the parents. This suggested that pairs of contrasting traits behave independently from each other and are inherited independently.

19. A man with blood group A marries a woman with blood group O. Their daughter has blood group O. Does this information allow you to determine whether the A or O trait is dominant? If yes, why? If not, why not?

Answer:

(i) If A is considered dominant over O, the following combinations are possible:

Man: A/O and Woman: O/O

18. How do Mendel's experiments show that traits are inherited independently?

Answer: Mendel crossed two pairs of contrasting traits in pea plants. The first generation (F1) showed only the dominant traits from each pair. In the second generation (F2), the offspring displayed both parental traits as well as new trait combinations not present in the parents. This suggested that pairs of contrasting traits behave independently from each other and are inherited independently.

19. A man with blood group A marries a woman with blood group O. Their daughter has blood group O. Does this information allow you to determine whether the A or O trait is dominant? If yes, why? If not, why not?

Answer:

(i) If A is considered dominant over O, the following combinations are possible:

Man: A/O and Woman: O/O

	A	O
O	AO	OO
O	AO	OO

Since the daughter has blood group O, it indicates that the man has the genotype A/O, meaning O is recessive to A.

	A	A
O	OA	OA
O	OA	OA
O	OA	OA
A	AA	AA

Thus, it is not possible to determine which blood group is dominant based on the information provided in the question.

20. What are the various ways in which organisms with specific traits increase in a population?

Answer: When a change occurs in a population, and that change contributes to the better survival of organisms in existing natural conditions, the trait is then naturally selected. Consequently, more organisms possessing that trait survive in the population.

21. Why are the traits acquired by an organism during its lifetime not heritable?

Answer: The traits acquired during an organism's lifetime are changes in the somatic cells and are not capable of being passed on to the next generation. Only changes that occur in the DNA of germ cells can be heritable.

22. Why is the survival of a small number of tigers a concern from a conservation perspective?

Answer: Tigers have very little variation in their genetic traits, and if natural conditions change drastically, any tiger may struggle to survive. For example, if a deadly disease infects tigers, the lack of genetic diversity in the population means that all tigers could potentially die. The decline in the tiger population suggests that the existing variations may not be suited to the changing environment and could soon face extinction.

23. Is geographic separation a significant factor in the classification of plant species that undergo self-pollination? If yes, why, and if no, why not?

Answer: No, since the population is self-pollinating, it is less influenced by factors resulting from geographic separation. In contrast, geographic separation is a significant factor in cross-pollinated species because it leads to rapid accumulation of differences between two geographically separated organisms.

24. Is geographic separation a significant factor in the classification of asexual reproducing organisms? If yes, why, and if no, why not?

Answer: Any organism that reproduces asexually does not have significant variations among its generations. Any changes caused by geographic separation cannot be passed on to future generations because these variations in DNA are not sufficient to create new species.

26. Why can all human beings, despite having different sizes, colors, and shapes, be said to belong to the same species?

Answer: Although there may be slight differences in the genetic makeup of individuals from different ethnic groups, there is no reproductive isolation. Reproductive isolation separates one species from another. Humans who differ in size, color, and appearance can interbreed and produce viable offspring.

Chapter: Our Environment

1. A good Aquarium is described as \_\_\_\_\_

- a) Biome
- b) Ecosystem
- c) Community
- d) Biosphere

Answer: b) Ecosystem

2. The percentage of energy released into the environment in the form of biological activity and heat by primary consumers is \_\_\_\_\_.

- a) 60%
- b) 70%
- c) 80%
- d) 90%

Answer: d) 90%

3. The ozone layer is essential because it absorbs \_\_\_\_\_  
a) Ultraviolet radiation b) Heat c) Solar radiation d) Infrared radiation

Answer: a) ultraviolet radiation

4. Waste that is not subjected to biological degradation is called \_\_\_\_\_

a) Organic waste b) Manure c) Plastic d) Kitchen waste

Answer: c) Plastic

5. The factor responsible for the destruction of the ozone layer is \_\_\_\_\_

a) Nitric acid b) CFC c) Carbon d) Oxygen

Answer: b) CFC

6. The radiation that splits oxygen molecules in the atmosphere into atoms is \_\_\_\_\_

a) Beta radiation b) Gamma radiation c) Alpha radiation d) Ultraviolet radiation

Answer: d) Ultraviolet radiation

7. In the sequence of grass → grasshopper → frog → snake → eagle, the producers are \_\_\_\_\_

a) Grass b) Grasshopper c) Frog d) Snake

Answer: a) Grass

8. If grass provides 5000 kJ of energy in the food chain of grass → deer → tiger, the amount of energy the tiger can obtain is \_\_\_\_\_.

- a) 5000 J   b) 500 J   c) 50 J   d) 50,000 J

Answer: c) 50 J

9. The chemical substance used in fire extinguishers is \_\_\_\_\_.

- a) Nitrogen   b) Oxygen   c) Carbon dioxide   d) Hydrogen

Answer: c) Carbon dioxide

10. The gas produced from the incomplete combustion of fossil fuels is \_\_\_\_\_.

- a) Carbon dioxide   b) Carbon monoxide   c) Nitrogen   d) Hydrogen

Answer: b) Carbon monoxide

11. The amount of energy that herbivores receive when they consume green plants is \_\_\_\_\_.

- a) 40%   b) 10%   c) 20%   d) 90%

Answer: b) 10%

12. The primary consumer in a food chain is \_\_\_\_\_.

- a) Tiger   b) Crocodile   c) Lion   d) Frog

Answer: b) Crocodile.

13. The component that blocks ultraviolet radiation from the sun is \_\_\_\_\_.

- a) Plants   b) Clouds   c) Ozone   d) Sea

Answer: c) Ozone



14. The disease seen in humans due to the depletion of ozone in the atmosphere is \_\_\_\_\_.

a) Diabetes b) Tuberculosis c) Cholera d) Skin cancer

Answer: d) Skin cancer

15. When a harmful toxin is transferred from one trophic level to another through food, it increases. This is called \_\_\_\_\_.

a) Pesticide b) Poisoning c) Biological magnification d) Toxic accumulation

Answer: c) Biological magnification.

16. Give two examples of human-made ecosystems \_\_\_\_\_

Answer: Gardens and ponds are examples of human-made ecosystems.

17. Name two decomposers in our ecosystem.

Answer: Bacteria and fungi.

18. Why are green plants called producers?

Answer: Because producers have the ability to capture solar energy through photosynthesis.

19. Name the acid-forming oxides that cause acid rain.

Answer: Sulfur oxides and nitrogen oxides.

20. Why do enzymes that breakdown carbohydrates not breakdown rocks?

Answer: The enzymes produced in the body perform specific functions. There are no enzymes in our body that can breakdown or decompose rocks, so it is not possible to decompose them.

21. What is the reason for any food chain not having more than four trophic levels?

Answer: Since only 10% of energy is transferred from one trophic level to the next, it becomes very limited as it moves from level to level, making it insufficient for more than four trophic levels.

**Diagrams to be Practiced for the 2024-25 Academic Year for Class 10 Students.**

SL No.	Chapter	Figure No	Figure Name
1.	Chemical Reactions and Equations	1.6	<ul style="list-style-type: none"><li>• Electrolysis of Water</li></ul>
2.	Acids, Bases and Salts	2.1	<ul style="list-style-type: none"><li>• Reaction of Zinc Granules with dilute Sulphuric Acid and Testing Hydrogen Gas by Burning</li></ul>
3.	Metals and Non-metals	3.3 3.12	<ul style="list-style-type: none"><li>• Action of steam on a Metal</li><li>• Electrolytic refining of Copper</li></ul>
4.	Life Processes	5.3 5.10 5.14	<ul style="list-style-type: none"><li>• a) Opened Stomata</li><li>• b) Closed Stomata</li><li>• Sectional view of Human Heart</li><li>• Structure of Nephron</li></ul>
5.	Control and Coordination	6.3	<ul style="list-style-type: none"><li>• The Human Brain</li></ul>
6.	How do organisms reproduce	7.8	<ul style="list-style-type: none"><li>• Germination of pollen on stigma</li></ul>
7.	Light – Reflection and Refraction	9.7	<ul style="list-style-type: none"><li>• Ray Diagrams of Images Formed by a Concave Mirror (a, b, c, d, e, and f)</li></ul>

		9.16	<ul style="list-style-type: none"> <li>• Nature, Position, and Size of Images Formed by a Convex Lens at Different Positions of Objects (a, b, c, d, e, and f)</li> </ul>
9.	The Human Eye and the Colourful World	10.6	<ul style="list-style-type: none"> <li>• Recombination of the Spectrum White Light</li> </ul>
10.	Electricity	11.1 11.1 Table 11.6 11.7	<ul style="list-style-type: none"> <li>• Simple Electric Circuit</li> <li>• Symbols commonly used in electrical circuits.</li> <li>• Resistors in series mode</li> <li>• Resistors in parallel</li> </ul>
11.	Magnetic Effects of Electric Current	12.6 (a)	<ul style="list-style-type: none"> <li>• A pattern of concentric circles indicating the field lines of a magnetic field around a straight conduction wire.</li> </ul>